KAURANE, ISOKAURANE, AND BEYERANE DERIVATIVES FROM PETERAVENIA SPECIES

E. Ellmauerer, J. Jakupovic, F. Bohlmann,

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, FRG

and R. SCOTT

Department of Botany, University of Texas at Austin, Austin, Texas 78713

ABSTRACT.—The investigation of two *Peteravenia* species afforded, in addition to known diterpenes, a new beyerene, two kaurane, and three isokaurene derivatives. The structures were elucidated by high-field ¹H-nmr spectroscopy. The chemotaxonomy of these species is discussed briefly.

The small genus *Peteravenia* (Compositae, tribe Eupatorieae) (1) is placed in the subtribe Hebecliniinae (2,3), a group of genera that are weakly differentiated from the Critoniinae and that were distinguished on the basis of their generally hirsute receptacles, elongated anther collars of subquadrate cells, and carpodia continuous with the lower parts of the achenes (2,3). These characters, however, are not very prounounced within *Peteravenia*. We have, therefore, studied two species chemically, and the results are discussed in this paper.

The aerial parts of *Peteravenia malvaefolia* (DC.) K. et R. afforded stigmasterol, β -sitosterol, and the diterpenes *ent*-16 β , 17-dihydroxykaurane [1] (4), *ent*-3 β -acetoxy-16 β , 17-dihydroxykaurane [2] (5), *ent*-3 β , 17-diacetoxy-16 β -hydroxykaurane [3], *ent*-kaur-15-en-17-oic acid [4], *ent*-3 β -acetoxy-kaur-15-en-17-oic acid [5], *ent*-3 β , 19-diacetoxy-kaur-15-en-17-oic acid [6], 3 α -hydroxybeyer-15-en-19-oic acid [7] (6), 18-acetoxybeyer-15-en-19-oic acid [8], and 3 α , 17-dihydroxybeyer-15-ene [9] (7). The acids 4-8 were isolated as the corresponding methyl esters 4a-8a.



The structures of 2 and 3 followed from their ¹H-nmr spectra (see Experimental section), which were close to that of 1 (4); the spectrum of 2 agreed with values reported for this acetate prepared from the corresponding ketone (5). As expected, the H-3 signal was shifted downfield in the diacetate 3 when compared with the chemical shift of H-3 in 2. The stereochemistry and also the position of the oxygen function of these derivatives were established by nOe difference spectroscopy. Clear nOe's were observed between H-5, H-3 (10%), and H-9 (4%), between H-19 and H-20 (5%), between H-18 and H-3 (12%) as well as between H-20, H-19 (7%), and H-14 (11%) (the first proton listed is always the one irradiated). Because the sign of the optical rotation of 2 was identical with that reported (5), the presence of *ent*-kauranes was established.

The structure of 4a also followed from its ¹H-nmr spectrum (see Experimental section). Most signals were close to those of **1** (4), but the changed situation in the fivemembered ring caused a downfield shift of the signal of H-13 and of H-15, which occurred as a singlet at δ 6.49; this can be explained by the presence of an isokaurene with a carbomethoxy group at C-16.

The ¹H-nmr spectra of **5a** and **6a** (see Experimental section) were close to that of **4a**; however, an additional lowfield signal in the spectrum of **5a** (δ 4.47 dd) and an acetate methyl singlet indicated the presence of a 3 α -acetoxy derivative of **4a**. In the spectrum of **6a**, pairs of doublets at δ 4.33 and 4.13 and a further acetate methyl singlet clearly showed that in **6a** one of the methyl groups was replaced by an acetoxy methylene group. The nOe difference spectroscopy indicated that a 19-acetoxy derivative of **5a** was present. Clear effects were observed between H-18, H-3 (12%), and H-19 (3%), between H-20, H-19 (5%), H-19' (7%), and H-14 (11%) as well as between H-3 and H-5 (5%). These effects clearly established the position of the acetoxy groups and also the stereochemistry at C-3—C-5 and C-8—C-10.

Compound 7a had been isolated previously (6). The proposed stereochemistry was established by nOe difference spectroscopy. Clear effects were obtained between H-20, OMe (2%), and H-15 (8%), between H-17, H-16 (6%), and H-14 as well as between H-3, H-5 (5%), and H-18 (6%). This technique also assigned the methyl singlets and the position of the oxygen functions. The ¹H-nmr spectrum of **8a** showed some similarities to that of 7a. However, the chemical shifts of one of the tertiary methyl groups differed clearly, and a second methyl singlet was replaced by a pair of doublets and an acetoxy methyl singlet. Furthermore, the H-3 signal was shifted upfield and was not separated from the upfield multiplets. Accordingly, the 3-hydroxy group was replaced by an 18-acetoxy group as required by the chemical shifts of H-19 and H-20. The spectral data of **9a** agreed with those reported in the literature (7). The ¹H-nmr data were in part similar to those of **7a** and **8a**, respectively. The absolute configurations of **4-9** were not determined, but biogenetic considerations favor the ones proposed.

The aerial parts of *Peteravenia schultzii* (Schnitts.) K. et R. gave germacrene D, stigmasterol, β -sitosterol, and the diterpenes **6**, **7**, and **8**. Previously, the investigation of a sample from Guatemala gave dehydronerolidol derivatives, while diterpenes were not observed (8).

The genus *Peteravenia* has been distinguished on the basis of its deciduous pappus setae, discolored phyllaries, and cordate leaves (1). The genus was said to be Critonioid due to its smooth corolla lobes, usually simple style base, and Ageratinoid in the placement of the embryo in the achene (1). The chemistry of the genus *Peteravenia* shows no relationship to *Decachaeta*, where sesquiterpene lactones are reported (9), or to *Hebeclinium* (10) both of which are placed in the Hebecliniinae. In the subtribe Critoniinae several genera contain diterpenes, mostly labdane derivatives (11,12); however, sesquiterpene lactones are also common (13-15). The accumulation of diterpenes of types 1-9 so far has not been reported from genera related to *Peteravenia*. More species from the two subtribes must be investigated to get a clearer picture.

EXPERIMENTAL

The air-dried plant material was extracted with MeOH-Et2O-petrol (1:1:1) at room temperature and separated following the methods reported previously (16). The extract of P. malvaefolia (240 g, collected in Nuevo Leon, Mexico, voucher B.L. Turner and co-workers 15611, stored in the herbarium of The University of Texas), after defatting with MeOH, was separated by cc (SiO₂) into four crude fractions: (a) Et₂Opetrol (1:3), (b) Et₂O-petrol (1:1), (c) Et₂O, and (d) Et₂O-MeOH (9:1) (all contained acids, and therefore, they were treated with CH2N2 in Et2O). Preparative tlc provided the following results: fraction a (SiO2, PF 254, Et₂O-petrol, 1:9) gave 166 mg of **4a**; fraction b (Et₂O-petrol, 3:7, two developments) gave 79 mg 4a, 20 mg 8a, 10 mg stigmasterol, and 10 mg β-sitosterol; fraction c (Et₂O) gave three bands (3/1-3/3). Preparative tlc of the bands gave the following: band 3/1 (Et₂O-petrol, 3:7, two developments) 31 mg 4a, 14 mg 8a, and 8 mg 5a; band 3/2 (Et₂O-petrol, 1:1, two developments) 74 mg 4a and 47 mg 5a. Hplc (RP 8, MeOH-H₂O, 4:1, ca. 100 bar) gave 12 mg 2 (R_t 5.0 min), 3.4 mg 9 (R_t 5.5 min), 9.6 mg 3 (R_t 7.4 min), and 2 mg 1 (R, 16.5 min). Preparative tlc of fraction d (Et₂O-petrol, 7:3) gave 67 mg 4a, 25 mg 5a, and two mixtures (4/3 and 4/4). Preparative tlc and hplc (RP 8, MeOH-H₂O, 4:1) of 4/3 gave 12 mg 5a and 2 mg 7a (R, 13.6 min). Hplc of 4/4 (RP 8, MeOH-H₂O, 4:1) gave 5 mg 6a (R, 7.8 min). The extract of P. schultzii (650 g, collected near Oaxaca, Mexico, voucher R. Scott A-111) gave, after addition of CH₂N₂ by cc and preparative tlc 10 mg 6a, 18 mg 7a, and 63 mg 8a. Known compounds were identified by comparing the 400 MHz ¹H-nmr spectra with those of authentic material.

ENT-3β,17-DIACETOXY-16β-HYDROXY-ENT-KAURANE [**3**].—Colorless oil; ir ν (CC1₄) cm⁻¹ 3600 (OH), 1750 (OAc); ms *m*/z 406.272 (M⁺) (0.7) (calcd for C₂₄H₃₈O₅: 406.272), 388 (M-H₂O) (2), 333 (M-CH₂OAc) (58), 273 (333-HOAc) (44), 135 (100); ¹H nmr (CDCl₃) 1.80 (ddd, H-1, *J*=13, 3.5, 3.5 Hz), 0.95 (ddd, H-1β, *J*=13, 13, 5.5 Hz), 1.68 (m, H-2), 4.45 (dd, H-3, *J*=10.5, 6 Hz), 0.84 (dd, H-5, *J*=13, 2 Hz), 1.37 (dq, H-6, *J*=3, 13 Hz), 1.53 (m, H-6β), 1.00 (d, H-9, *J*=6.5 Hz), 2.05 (br s, H-13), 1.93 (d, H-14), 1.64 (m, H-14'), 4.22 (s, H-17), 0.85 (s, H-19), 1.04 (s, H-20); [α] ²⁴D -19° (CHCl₃, c 0.25).

METHYL-ENT-KAUR-15-EN-17-OATE [4a].—Colorless crystals, mp 138°; ir ν (CCl₄) cm⁻¹ 1705, 1620 (C=CCO₂R); ms *m*/z 316.240 (M⁺) (90) (calcd for C₂₁H₃₂O₂: 316.240), 301 (M-Me) (100), 285 (M-OMe) (8), 193 (57), 192 (63), 191 (46), 81 (96); ¹H nmr (CDCl₃) 2.93 (br s, H-13), 2.16 (d, H-14, *J*=10.5 Hz), 6.49 (s, H-15), 0.87 and 0.89 (s, H-18, H-19), 1.06 (s, H-20), 3.72 (s, OMe); $[\alpha]^{24}D$ -29° (CHCl₃, *c* 1.6).

METHYL-ENT-3β-ACETOXY-KAUR-15-EN-17-OATE [**5a**].—Colorless oil; ir ν (CCl₄) cm⁻¹ 3540 (OH), 1730, 1250 (OAc), 1705, 1620 (C=CCO₂R); ms m/z 374.246 (M⁺) (15) (calcd for C₂₃H₃₄O₄: 374.246), 314 (M-HOAc) (70), 299 (314-Me) (16), 61 (100); ¹H nmr (CDCl₃) 4.47 (dd, H-3, J=11, 5.5 Hz), 2.93 (br s, H-13), 2.13 (d, H-14, J=10.5 Hz), 6.49 (s, H-15), 0.84 and 0.87 (s, H-18, H-19), 1.08 (s, H-19), 3.73 (s, OMe); [α]²⁴D -43° (CHCl₃, c 1.0).

METHYL-ENT-3β, 19-DIACETOXY-KAUR-15-EN-17-OATE [**6a**].—Colorless oil; ir ν (CCl₄) cm⁻¹ 3540 (OH), 1745, 1250 (OAc), 1720, 1630 (C=CCO₂R); ms m/z 432.251 (M⁺) (5.5) (calcd for C₂₅H₃₆O₆: 432.251), 372 (M-HOAc) (10), 312 (372-HOAc) (52), 297 (312-Me) (22), 61 (100); ¹H nmr (CDCl₃) 4.55 (dd, H-3, J=10, 8 Hz), 0.99 (d (br), H-5, J=13 Hz), 1.92 (ddd, H-6, J=13.5, 4.5, 3.5 Hz), 2.95 (br s, H-13), 2.10 (d, H-14, J=10.5 Hz), 6.48 (s, H-15), 1.01 (s, H-17), 4.33 and 4.13 (d, H-19, J=12 Hz), 1.10 (s, H-20), 3.73 (s, OMe).

METHYL-18-ACETOXYBEYER-15-EN-19-OATE [**8a**].—Colorless oil; ir ν (CCl₄) cm⁻¹ 1750, 1255 (OAc), 1740 (CO₂R); ms m/z 374.246 (M⁺) (36) (calcd for C₂₃H₃₄O₄: 374.246), 342 (M-MeOH) (4), 314 (M-HOAc) (20), 299 (314-Me) (3), 254 (314-HCO₂Me) (10), 55 (100); ¹H nmr (CDCl₃) 2.29 (d (br), H-3, J=13.5 Hz), 0.88 (ddd, H-3β, J=13, 13, 5 Hz), 5.68 (d, H-15), 5.45 (d, H-16, J=5.5 Hz), 0.99 (s, H-17), 3.92 and 4.39 (d, H-19, J=10.5 Hz), 0.59 (s, H-20), 3.67 (s, OMe), 2.02 (s, OAc); [α]²⁴D -9° (CHCl₃, c 0.77).

LITERATURE CITED

- 1. R.M. King and H. Robinson, Phytologia, 21, 394 (1971).
- H. Robinson and R.M. King in: "The Biology and Chemistry of the Compositae." Ed. by V.H. Heywood, J.B. Harborne, and B.L. Turner, Academic Press, London, 1977, p. 467.
- 3. R.M. King and H. Robinson, Phytologia, 46, 446 (1980).
- 4. F. Bohlmann, J. Jakupovic, H. Robinson, and R.M. King, Phytochemistry, 19, 2769 (1980).
- 5. J.R. Hanson and A.F. White. Tetrahedron, 26, 4839 (1970).

- 6. F. Bohlmann and P. Wegner, Phytochemistry, 21, 1175 (1982).
- 7. H.A. Lloyd and H.M. Fales, Tetrabedron Lett., 4891 (1967).
- 8. F. Bohlmann and A. Suwita, Phytochemistry, 17, 567 (1978).
- 9. V. Castro, F. Ciccio, S. Alvarado, F. Bohlmann, G. Schmeda-Hirschmann, and J. Jakupovic, Lieb. Ann. Chem., 974 (1983).
- 10. F. Bohlmann and M. Grenz, Chem. Ber., 110, 1327 (1977).
- 11. F. Bohlmann, C. Scheidges, R.M. King, and H. Robinson, Phytochemistry, 23, 1190 (1984).
- 12. F. Bohlmann, C. Zdero, R.M. King, and H. Robinson, Lieb. Ann. Chem., 2127 (1983).
- 13. F. Bohlmann, C. Zdero, and B.L. Turner, Phytochemistry, 24, 1263 (1985).
- 14. F. Bohlmann, C. Zdero, R.M. King, and H. Robinson, Phytochemistry, 23, 1187 (1984).
- 15. F. Bohlmann, R.N. Baruah, R.M. King, and H. Robinson, Planta Med., 167 (1985).
- 16. F. Bohlmann, C. Zdero, R.M. King, and H. Robinson, Phytochemistry, 23, 1979 (1984).

Received 15 August 1986